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## Recent developments in N-fuels, $NO_x$ and $NO_x$ /hydrocarbons interaction chemistry

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The investigation of nitrogen chemistry is gaining extreme attention in relation to sustainable energy technologies and clean combustion. Beside the long-lasting interest in nitrogen oxides  $(NO_x)$  chemistry for pollutants emission mitigation and abatement, additional challenges are outlined by the imminent needs of the energy transition.

(1) Ammonia (NH<sub>3</sub>) is a promising carbon neutral fuel of interest also for diesel fuel mixtures. (2) Biomass conversion to bio-oils (FPBO) with (fast) pyrolysis is a sustainable way to produce bio-fuels for transportation engines, heat and power applications. Despite the intrinsic carbon neutral nature, FPBO combustion carries some concern in relation to the impact of fuel-N on NO<sub>x</sub> emissions. (3) Demand for fuel-flexible combustion systems (e.g. NH<sub>3</sub>/Diesel, NH<sub>3</sub>/H<sub>2</sub>/CH<sub>4</sub>), (4) optimization or retrofitting of consolidated technologies (EGR), and (5) design of new processes (e.g. MILD combustion) require improved understanding of fuel/NO<sub>x</sub> interactions. From a microscopic chemical kinetics perspective, all of the above share a common core set of relevant N-species and reactions, making the development of a physically meaningful, consistent and single model mandatory.

This work presents the recent developments in the N-fuels (NH<sub>3</sub>, pyrrole, pyrrolidine), NO<sub>x</sub> and NO<sub>x</sub>/hydrocarbons subsets of the CRECK kinetic model. From the molecular scale calculation of rate constants and thermochemistry, criticisms and model development/optimization strategies for such interconnected subsets are discussed, together with applications in more complex reactor-scale simulations.